New insights into colloidal phase transitions using neutron scattering techniques

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INTRODUCTION

- Fundamentals of crystallisation and glass transition are not fully understood.
- Binary colloidal suspensions are a good model.
- Brownian, real-time motion.
- Limited studies on dynamics and structure. [1-3]

RESEARCH QUESTIONS

- 1. What combination of nanoparticle and **solvent** will be the ideal colloidal system to study?
- 2. How do different volume fractions ϕ affect the crystallisation and phase behaviours of the system? **Volume**_{particle}

3. How does change in polydispersity affect the crystallisation and phase behaviours of the system?

Volume_{total}



4. What **resolution** do we need to measure the full range of scattering vectors q during solidification processes in our neutron scattering experiments? NST APPROV National Institute a Standards and Technology

CHOOSING THE IDEAL SYSTEM

Particle



- Large particles, relatively cheap, non-volatile. Testing: Silica (d = 20-2470 nm) and Polystyrene (d = 62-2500 nm) Solvent
- Refractive index **n** match, similar density ρ , transparent, relatively cheap, reasonable viscosity **n**. Potentials: ionic liquids and deep eutectic solvents.

Testing: **BMIm Acetate** and **BMIm Thiocyanate**





PRELIMINARY RESULTS Checking particles are stable with DLS.



Contact:







TECHNIQUES DLS & SEM, RMMF SAXS, RMMF & Australian Synchrotron



U/SANS, ANSTO



PREVIOUS WORK [4-8]



Fig. 3 SANS data for SMU39 at a volume fraction of 0.54 over a period of ~250 minutes following the quench. The inset highlights the growth of the main crystalline peak with Gaussian fits.

A. Brands, H. Versmold and W. van Megen, J. Chem. Phys., 110: 1283 (1999) 2. S.R. Williams and W. van Megen, Phys. Rev. E, 64: 041504 (2001). 3. P.C. Royall, A.A. Louis, H. Tanaka, J. Chem. Phys. 127: 044507 (2007). 054505 (2011).



VSANS, NIST



REFERENCES

4. W. van Megen, V.A. Martinez, and **G. Bryant**, Phys. Rev. Lett. 102, 168301 (2009). 5. V.A. Martinez, G. Bryant and W. van Megen, Phys. Rev. Lett. 101: 135702 (2008). W. van Megen, V.A. Martinez and G. Bryant, Phys. Rev. Lett. 103: 258301 (2009). 7. V.A. Martinez, **G. Bryant** and W. van Megen, J. Chem. Phys. 133: 114906 (2010). V.A. Martinez, J.H.J. Thijssen, F. Zontone, W. van Megen and G. Bryant, J. Chem. Phys. 134: